



# Molecular Electronics

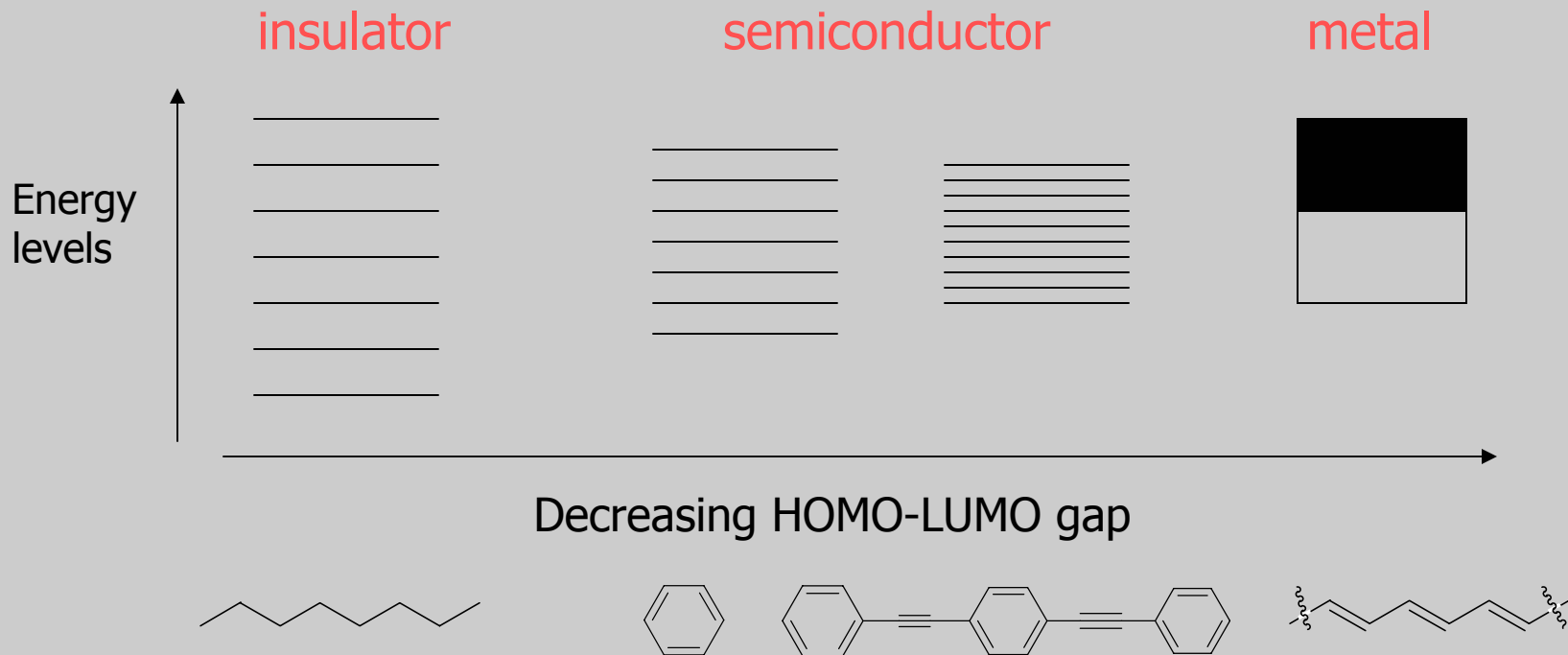


Dr. Bin Yu

Electrical Engineering Course 379

# Molecular Band Structure

- Molecules can be insulator, semiconductor and metal
- The band gap of a molecule can be tuned by its structure



# Brief History of Molecular Electronics

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1970



1980



1990



2000



2010



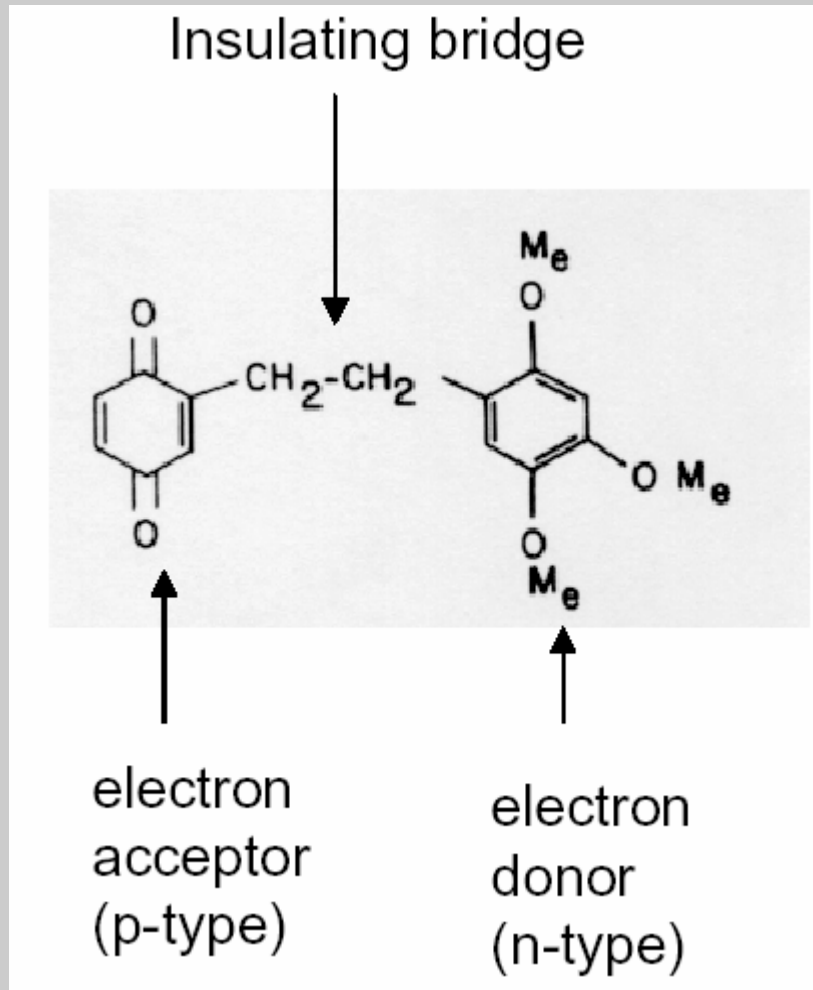
Molecular logic and  
memory device  
demonstrated

Spiro-switch synthesized  
Different variants of  
molecular rectifiers made  
Individual molecular  
conductance measured  
Modeling work

Sensitive new tools  
invented

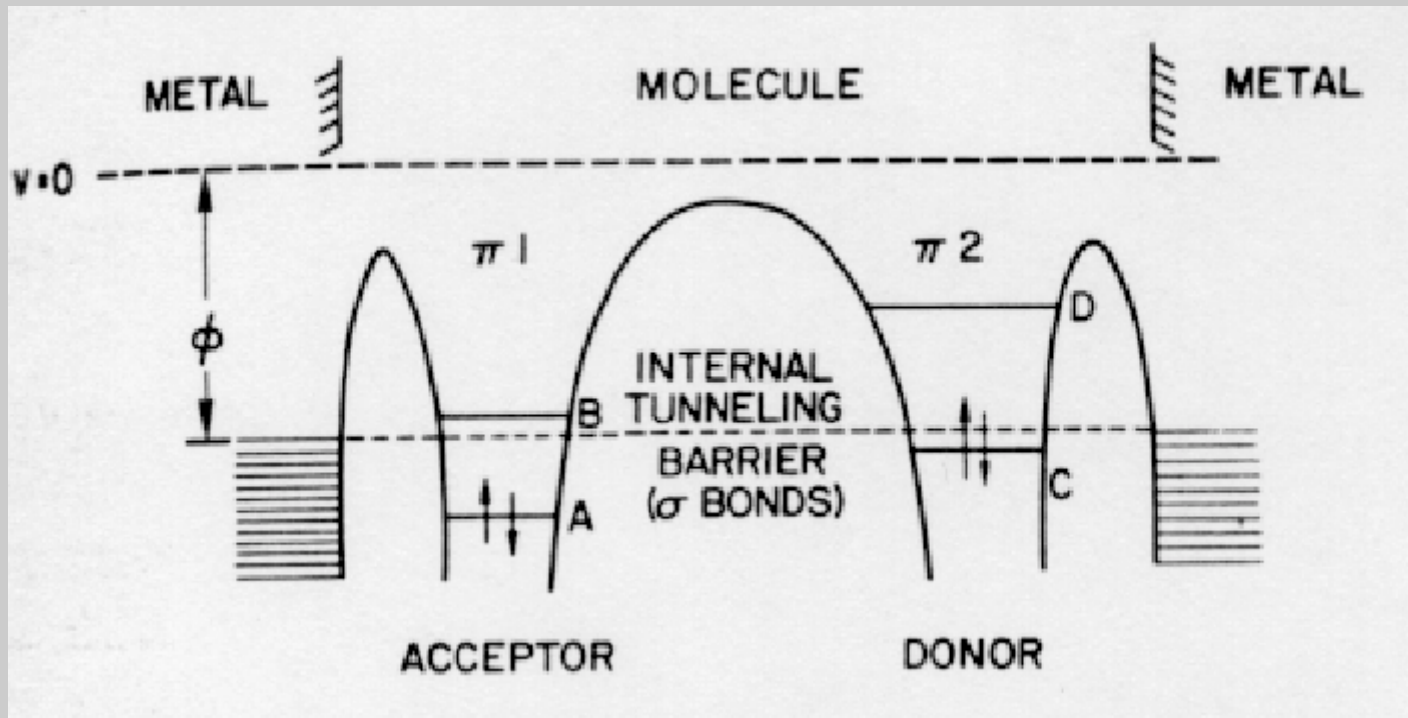
- Theory on molecular  
rectification

# The Beginning of Molecular Electronics

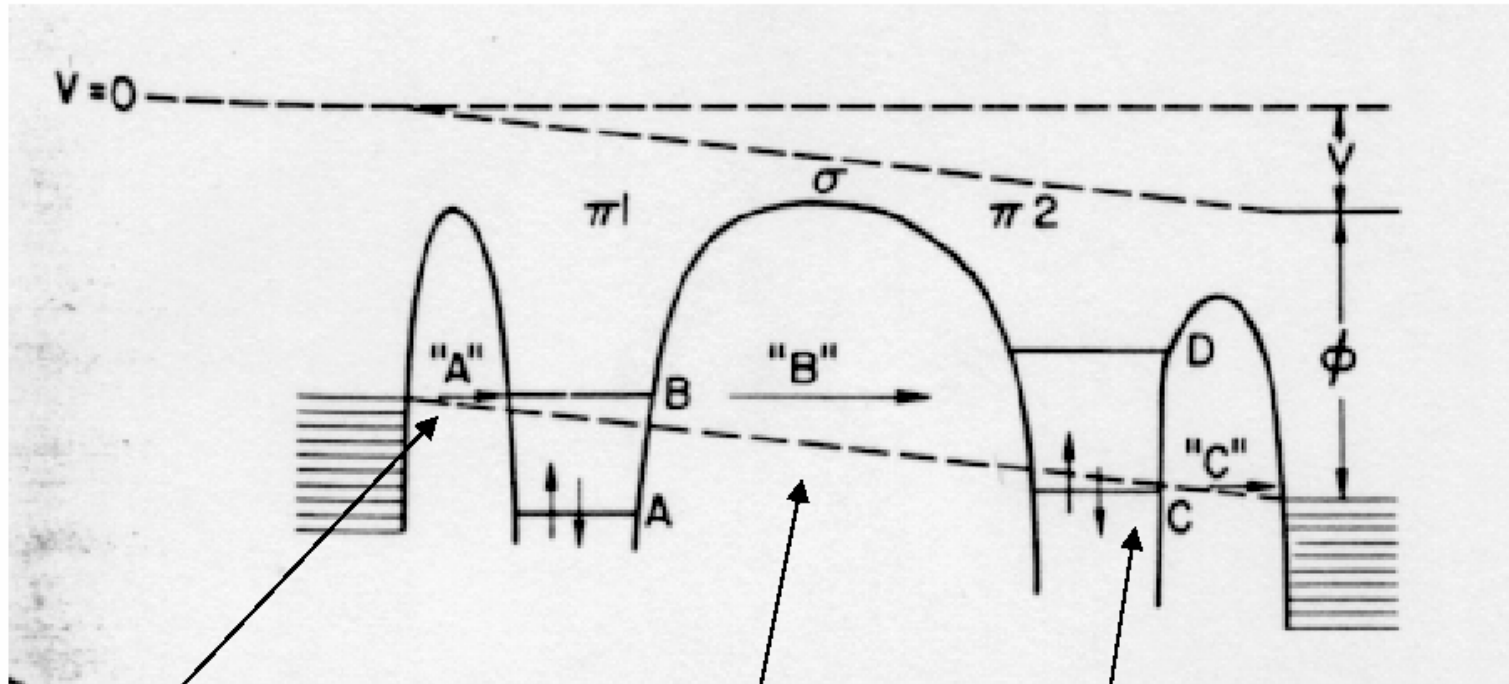


- In 1974 Aviram and Ratner predicted that molecules with a donor and acceptor would like a pn junction.

# A Donor-Acceptor Junction



# Forward Bias



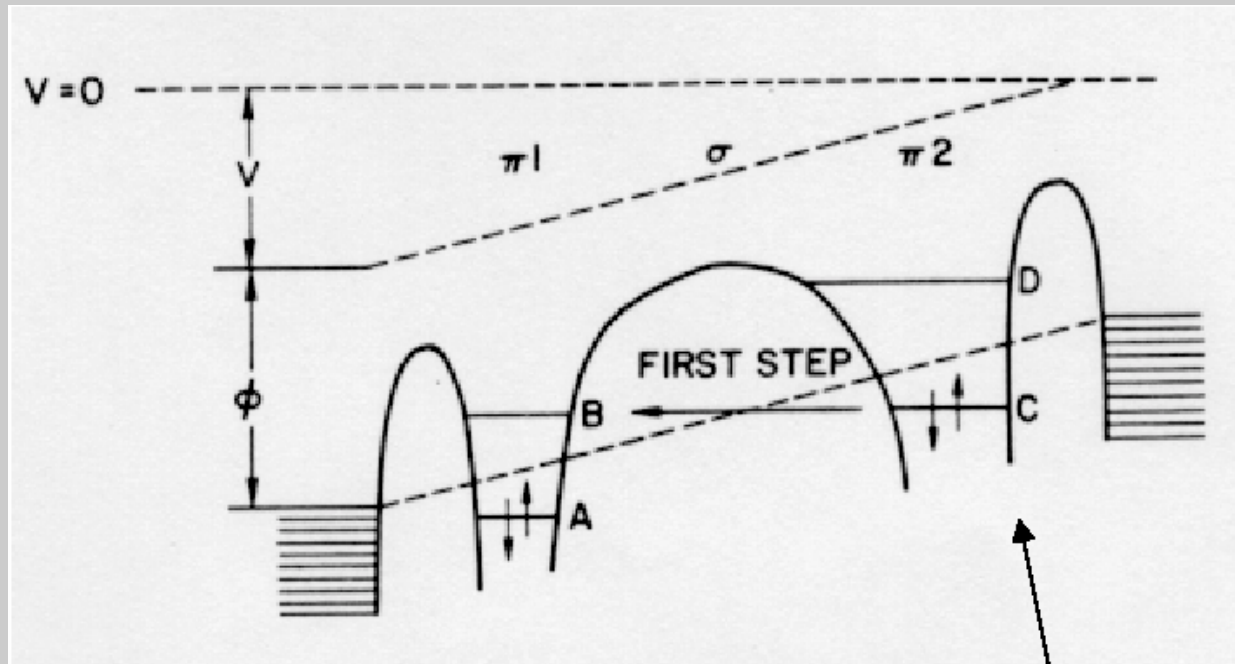
Electrons can easily tunnel across this barrier.

Holes can easily tunnel across this barrier.

Electrons can tunnel from B to C if there are holes in C.

Current flows.

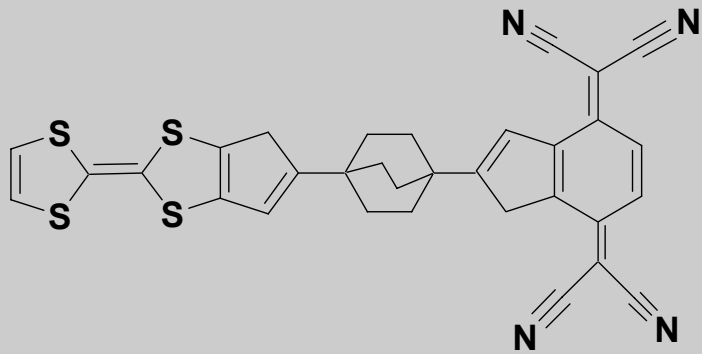
# Reverse Bias



Electrons can't tunnel into D.

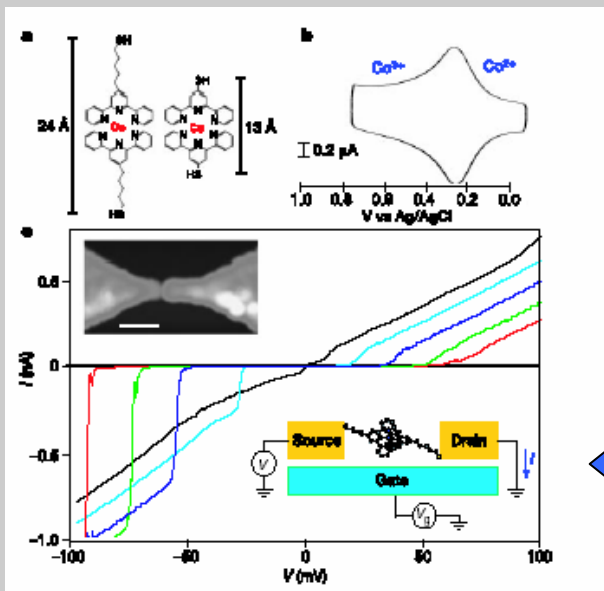
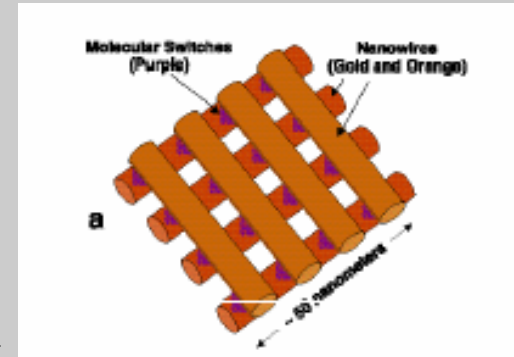
Very little current flows.

# Molecule: Active Electronic Component



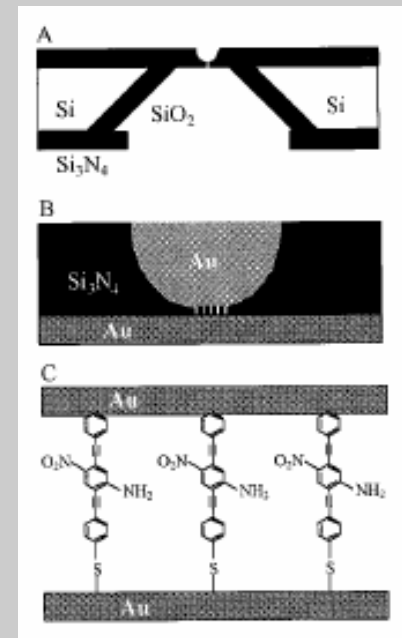
The Aviram-Ratner  
molecular rectifier  
1974

Memory crossbar  
HP, 2001



Nanopore  
Reed and Tour  
1999

Kondo effect  
McEuen and Park  
2002





# Why Molecule?

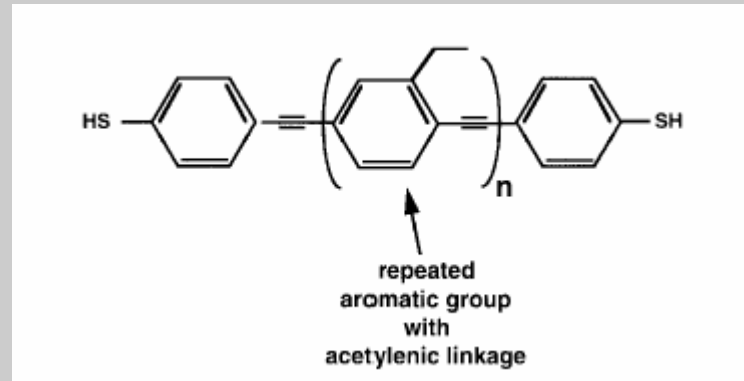
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- 2 decades of work
- Individual molecules or supramolecular structures
- Covalently bonded molecular structures, electrically isolated from bulk substrate
- Different from “organic devices” that use bulk-effect electron transport
- Easily made identical by the trillions of billions
- Great power and variety of organic chemistry offers more options for designing and fabricating nm-scale devices than Si
- Inexpensive

# Molecular Structures

- Molecular nanowires

- A sequence of benzene-like rings connected by acetylene linkage
- Orbital or clouds of  $\pi$ -electrons to form a single large orbital through the length of wire to permit mobile electrons to flow.



- Buckytubes

- Carbon nanotubes
  - 10nm diameter CNT => carry 10mA current

# Why Molecular Devices?

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- Nano-scale Dimension
  - The “last” engineering level human can handle
  - Few/single electron operation - low voltage / low power
- Uniform Components
  - Achieved by chemistry synthesis – less variation
- Tunable Functionality
  - Availability of many molecule species tailored for different device applications through surface chemistry engineering

Great potential to build **ultra-dense**, **low-power**, **low-cost** computing chips (logic, memory)

# Molecular Transport

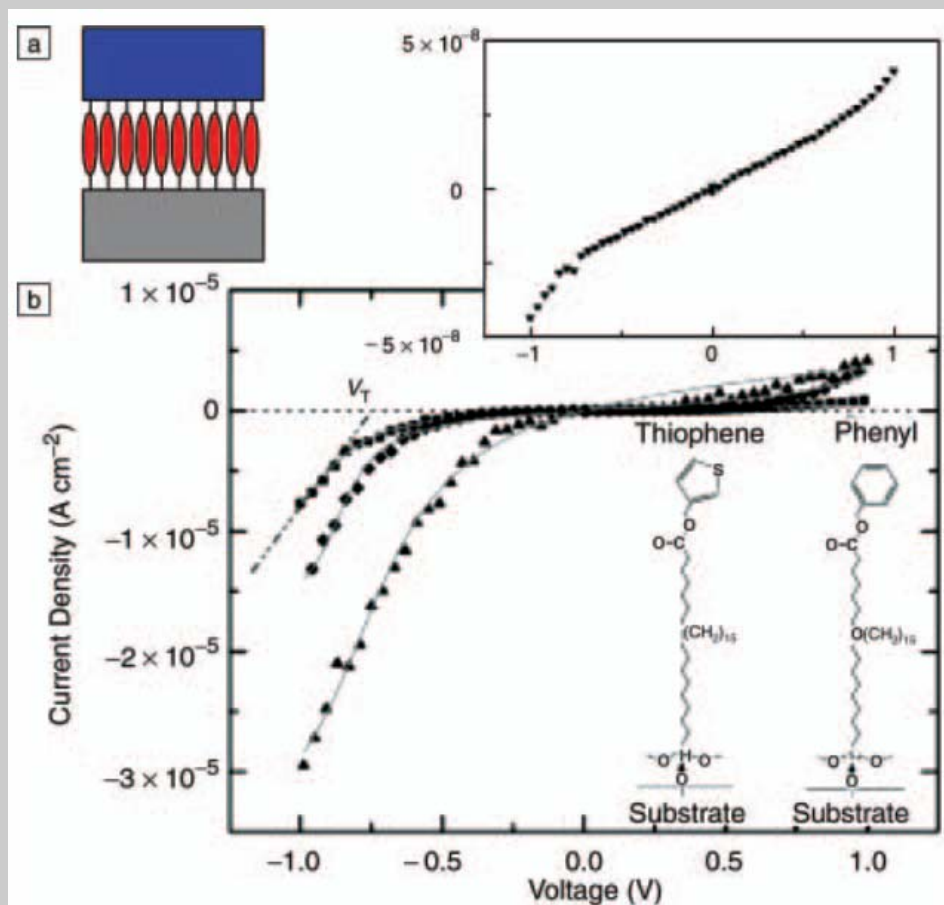
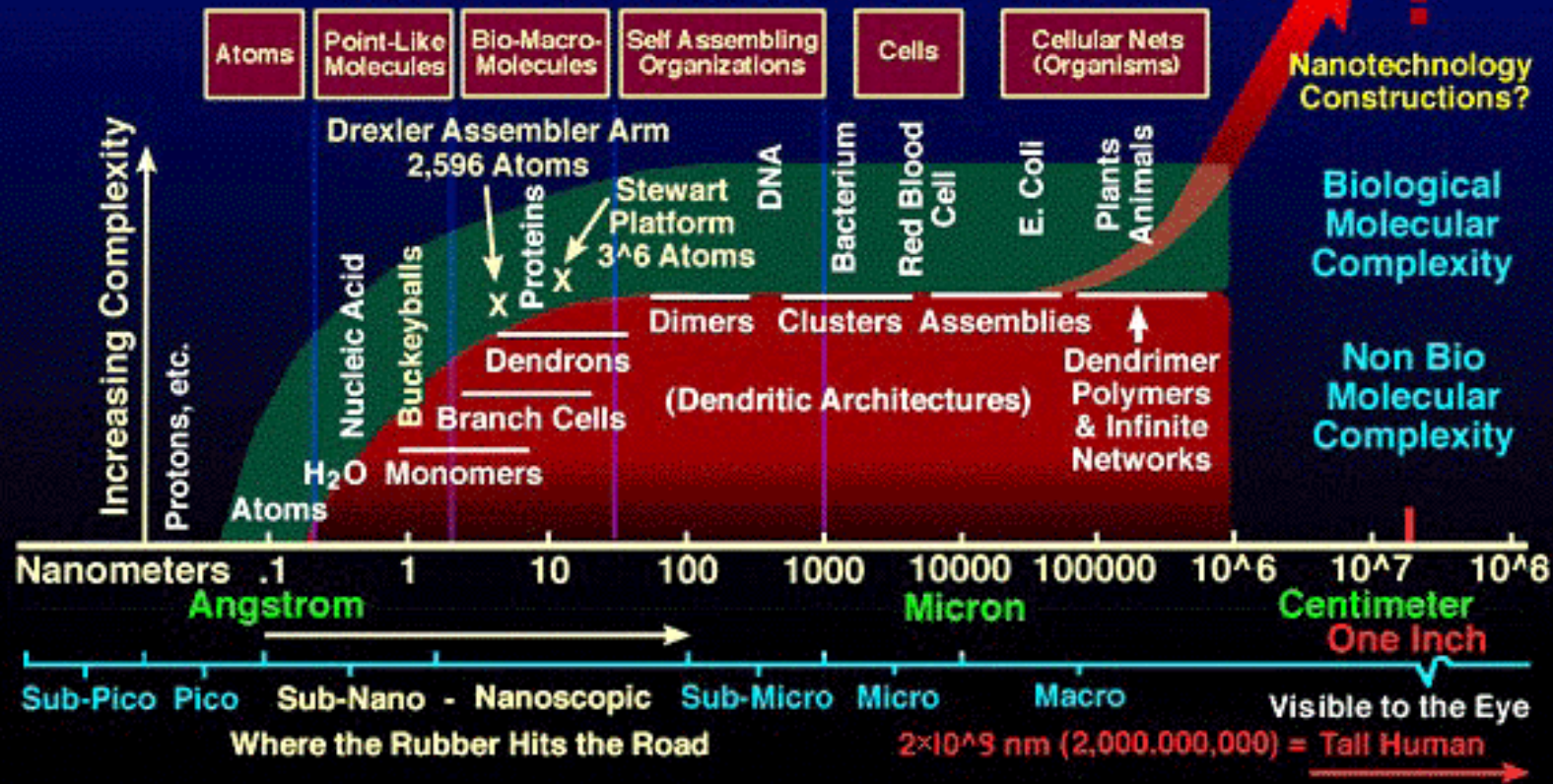
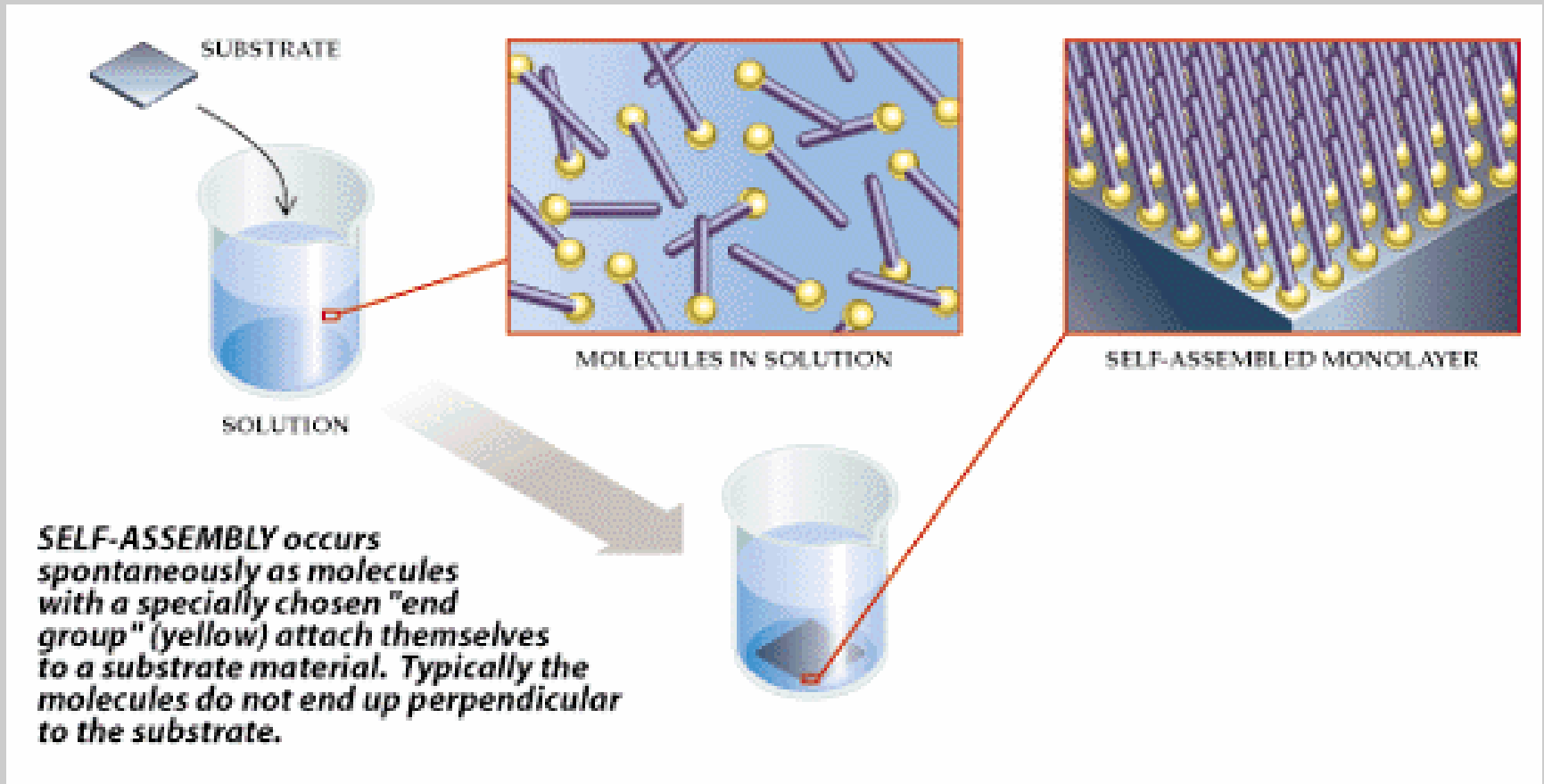


Figure 4. (a) Schematic illustration of a two-terminal vertically fabricated junction. (b) I-V characteristics showing rectification of a thiophene-terminated oct-7-en-1-trichlorosilane self-assembled monolayer (triangles), and phenyl-terminated (squares) and thiophene-terminated (diamonds) heptadec-16-en-1-trichlorosilane self-assembled monolayers, in a molecular junction between Al and n<sup>+</sup> Si electrodes. The top graph shows the graphical determination of the threshold voltage.  $V_T$  is taken as the intercept of the linear extrapolation of the I-V curve and the zero of the y axis. (Reprinted with permission from Reference 24.)

# Molecular Scale & Complexity

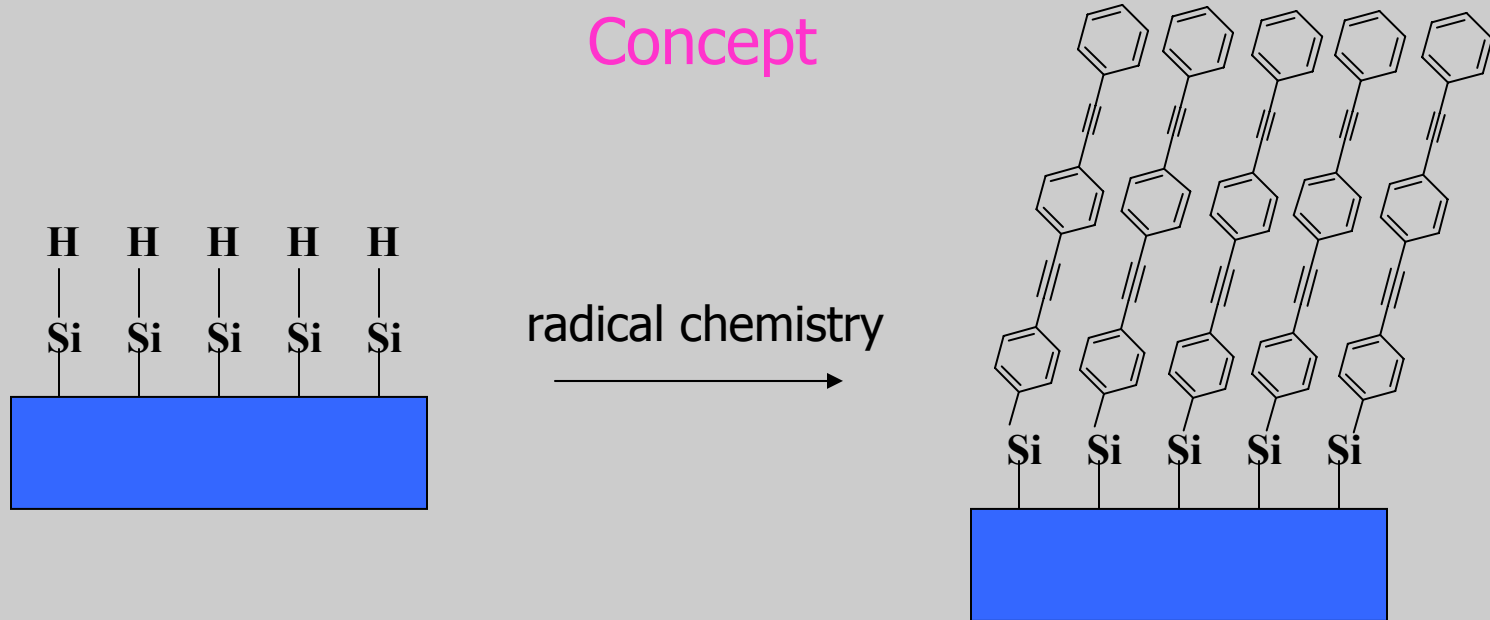


# Molecular Self-Assembly



- At NASA lab, functional organic molecular monolayers can be self-assembled onto metal (Au) and semiconductor (Si,  $\text{In}_2\text{O}_3$ ) surfaces

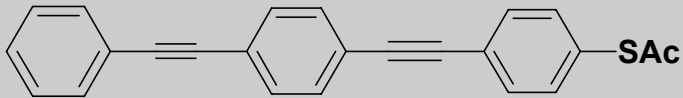
# Radical Chemistry/Direct Functionalization on Si or Ge Surface



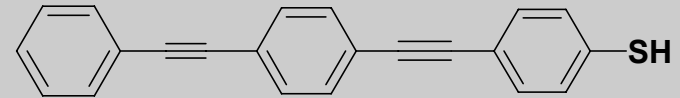
Direct attachment of conjugated systems onto Si surface will minimize coulomb blockade and give optimal electron transport behavior. In addition, the covalent nature of Si-C bond will give stable and robust devices.



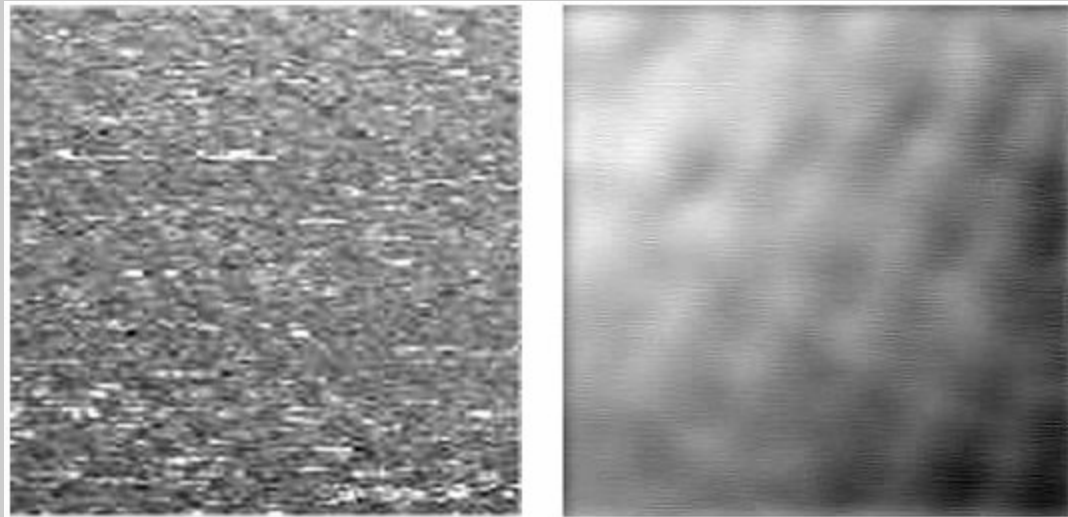
# Self-Assembled Molecular Monolayer



1. Ammonium hydroxide/  
tetrahydrofuran
2. Deposition on Au



1. Deposition on Au



The nature of the terminal group influences the order of the SAM.



# Molecular Design Basis

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## Device Requirement

- (1) Good self-assembly property
- (2) Make good contacts
- (3) Conducting
- (4) Switching at low potential
- (5) Magnetic property
- (6) Optical/opto-electronic property
- (7) Amenable to high density memory arrays
- (8) Robust and stable

## Chemical Synthesis

- (1) Rigid rod-like structure
- (2) Conjugated pi-electron system
- (3) Facile redox properties
- (4) Conformational barriers
- (5) Metal-ligand charge transfer
- (6) Transition metals with unpaired electrons
- (7) Imbedded chemical interactions at strategic and periodic positions in molecules

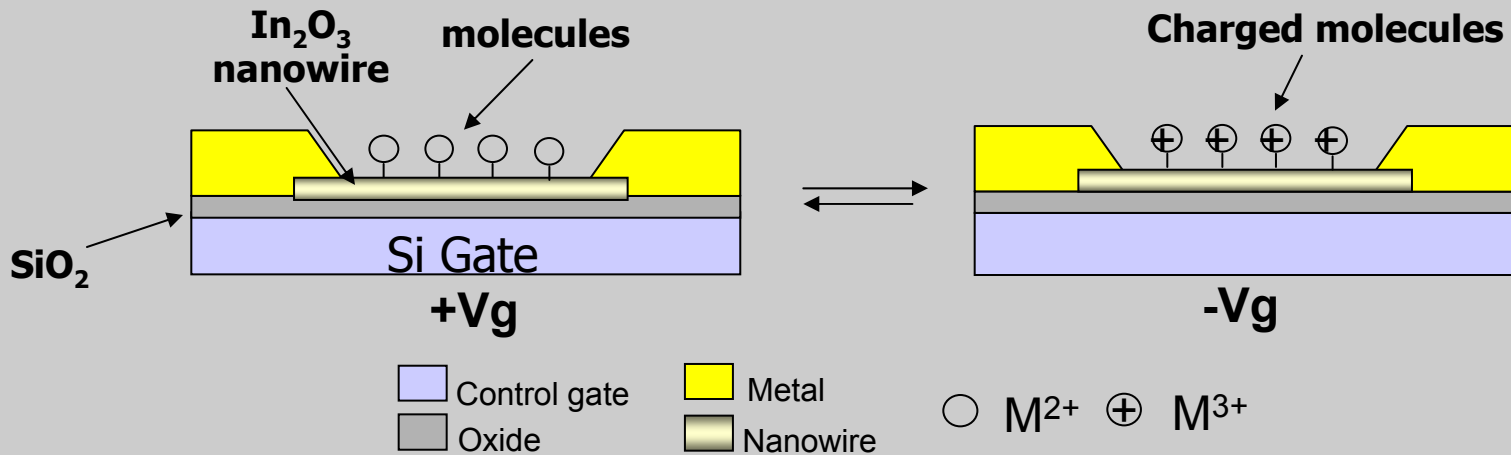
Band gap engineering –

# Molecular Memory: Issues

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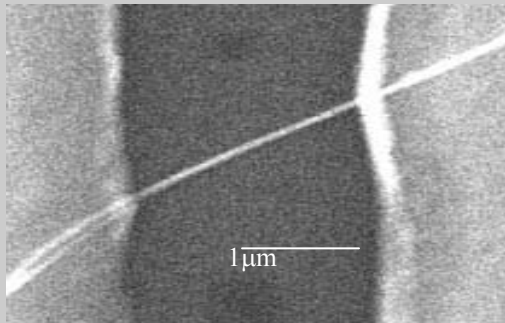
- Adhesion to ultra-small nanowire/nanotube for novel devices
- Localized charge distribution in the intermolecular direction – discrete charge storage nodes
- Molecule-semiconductor interface
  - Tunneling mechanism
  - Charge retention
  - endurance
- Quantum well effect – Coulomb Blockage?
- Thermal stability

# Molecular-Nanowire Memory



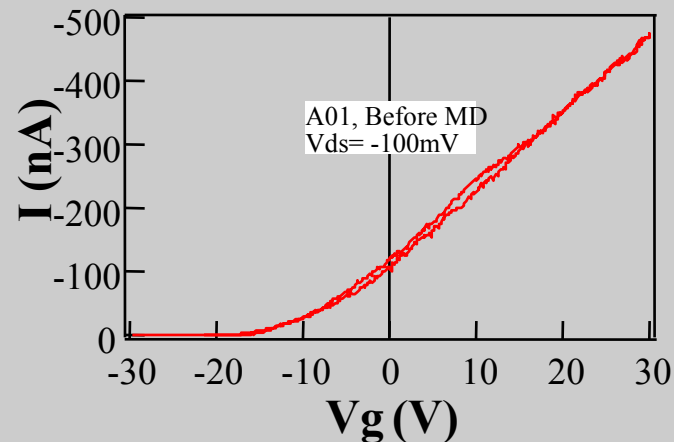
**Key feature: Molecules are chemisorbed on doped In<sub>2</sub>O<sub>3</sub> surface, molecular component (ligand/linker) replaces insulating oxide as charge transfer barrier.**

As-fabricated device



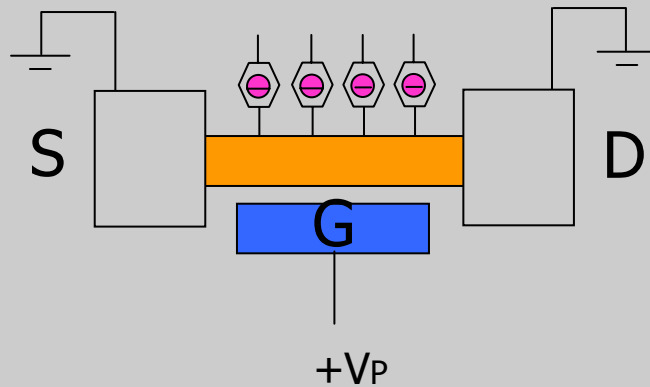
Li, et al. App. Phys. Lett. 2004.

I<sub>d</sub>-V<sub>g</sub>

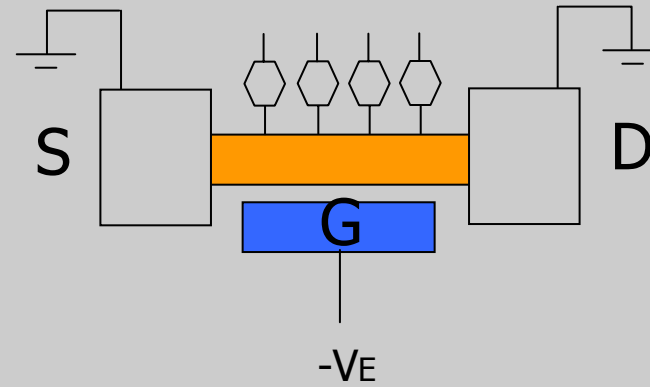


# Molecular Memory

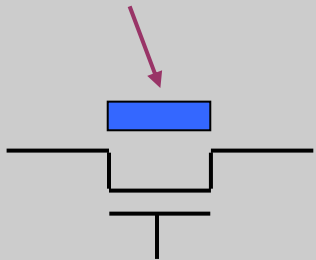
Programming (high  $V_{th}$ )



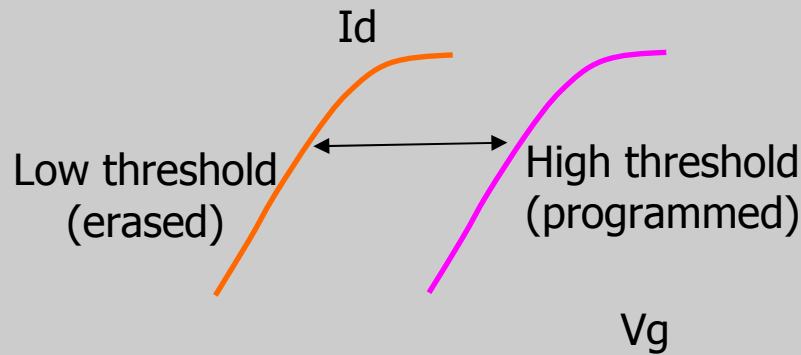
Erasing (low  $V_{th}$ )



charge storage material

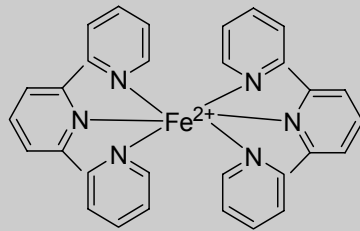


molecule memory  
(non-volatile)

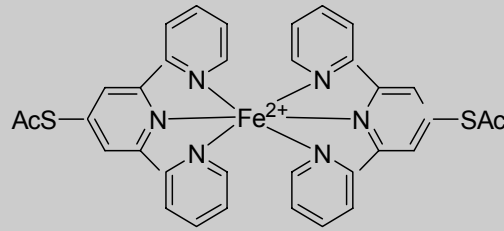


- Large gate bias for P/E
  - Positive  $+V_P$  for programming
  - Negative  $-V_E$  for erasure
- “Chemical gate” to modulate channel conductance

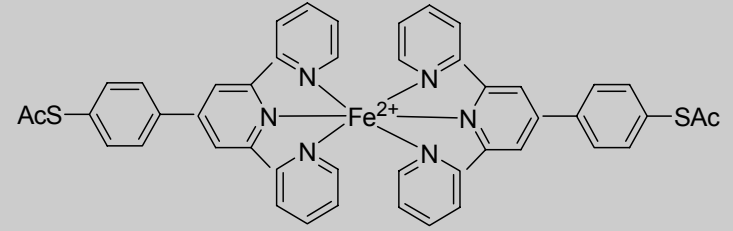
# Contact and Barrier vs. Retention Time



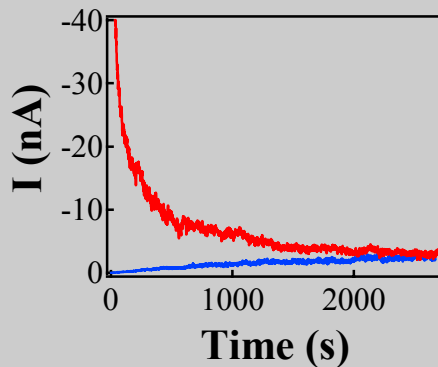
2Cl<sup>-</sup>



2Cl<sup>-</sup>

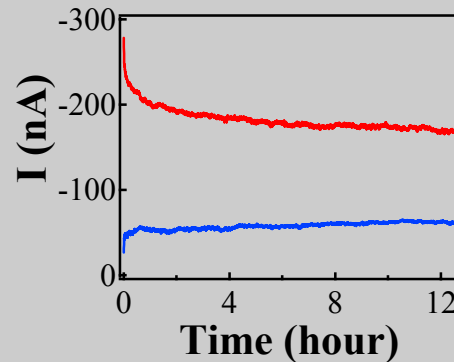


2Cl<sup>-</sup>



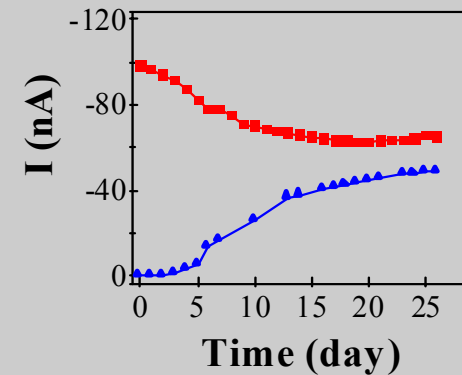
$\tau \sim 200$  seconds

Physisorption  
Poor contact  
Weak barrier



$\tau \sim 12$  hours

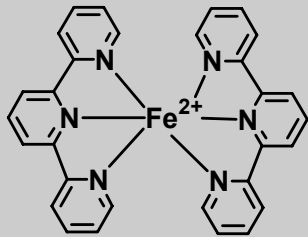
Chemisorption  
Robust contact  
Strong barrier



$\tau \sim 250$  hours

Chemisorption  
Robust contact  
Stronger barrier  
(Conformation)

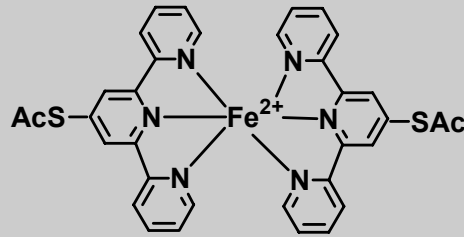
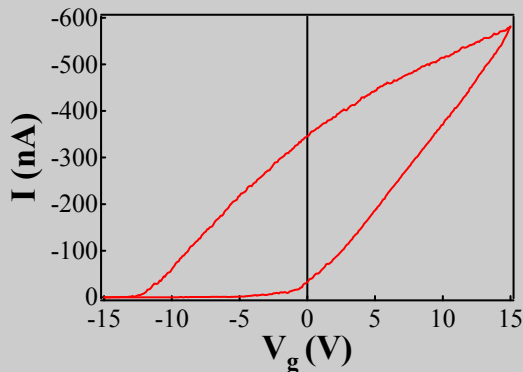
# Molecular Self Assembly on Nanowire



2 Cl<sup>-</sup>

**Physisorption**  
**Poor contact**  
**Weak barrier**

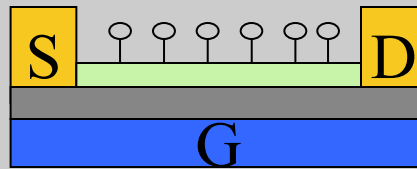
**$\tau_{on}, \tau_{off} \sim 100$  seconds**



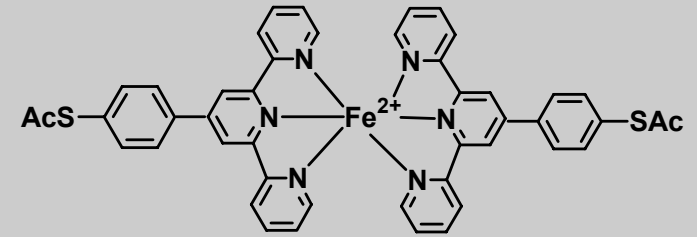
2 Cl<sup>-</sup>

**Chemisorption**  
**Robust contact**  
**Strong barrier**

**$\tau_{on}, \tau_{off} \sim 2-3$  hours**



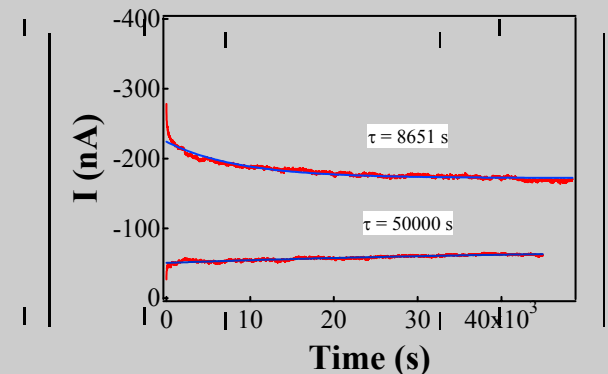
(W. Fan, NASA)



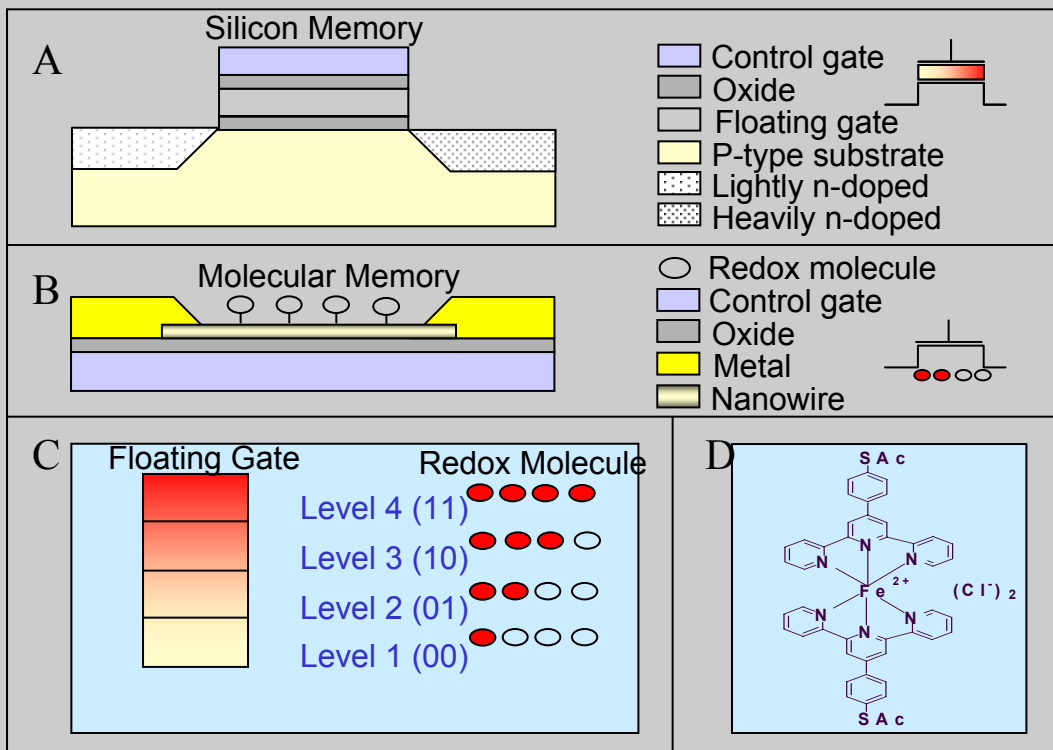
2 Cl<sup>-</sup>

**Chemisorption**  
**Robust contact**  
**Stronger barrier**  
**(Conformation)**

**$\tau_{on}, \tau_{off} \sim 600$  hours**

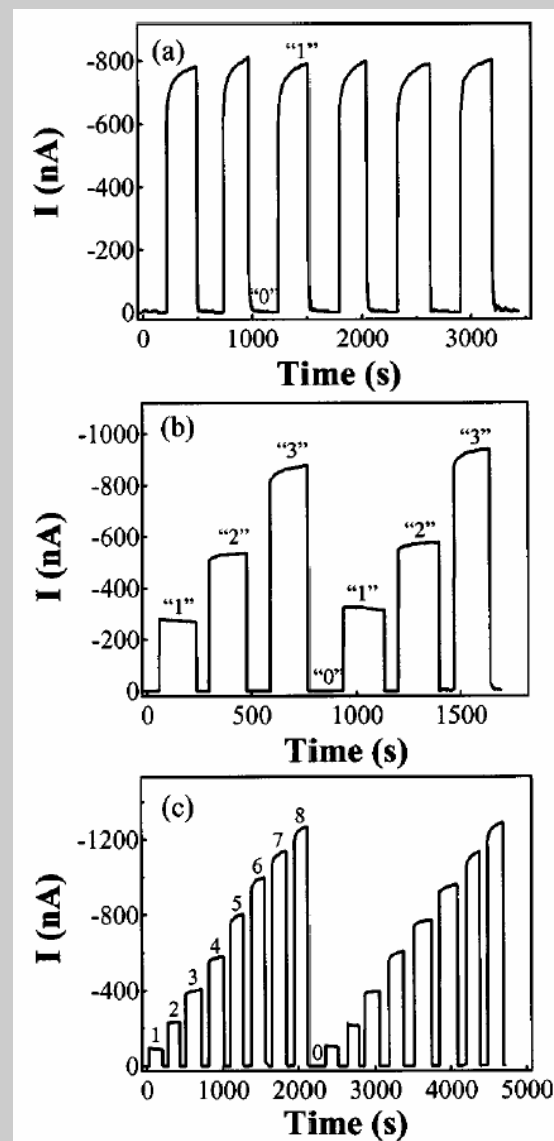


# Multi-Level Molecular Memory

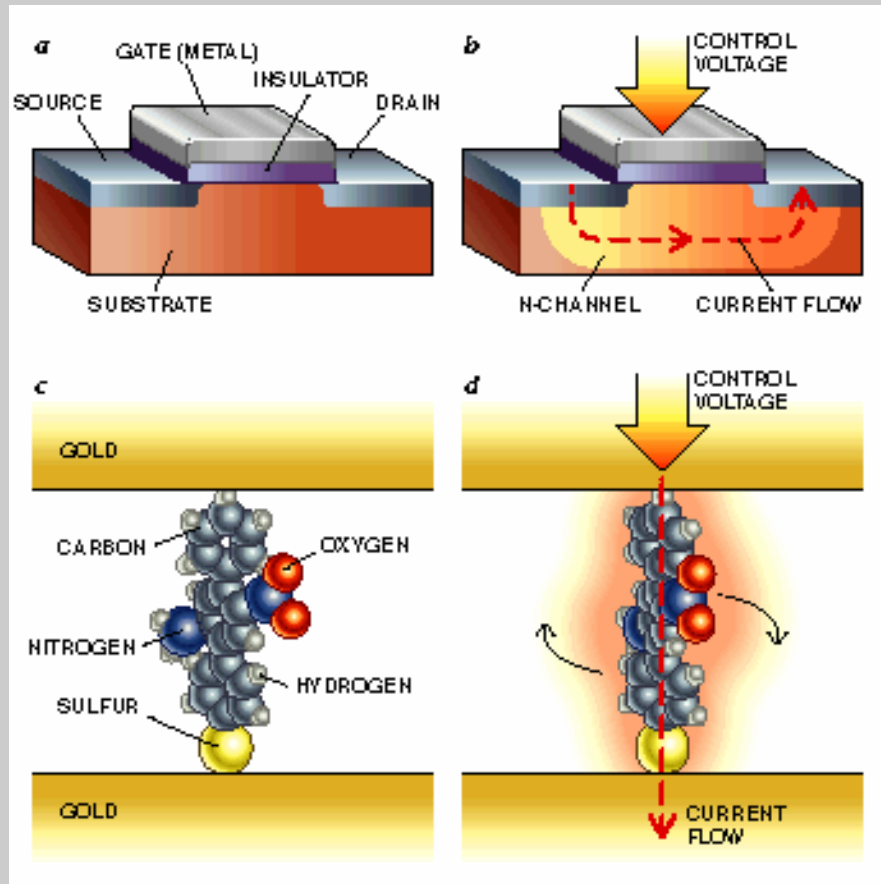


- Multilevel charge storage in redox molecules
- Nonvolatile nature (charge retention ~ 600 hours)

(NASA/USC, APL, March 2004)



# Molecular Switch



A molecule based on three benzene rings

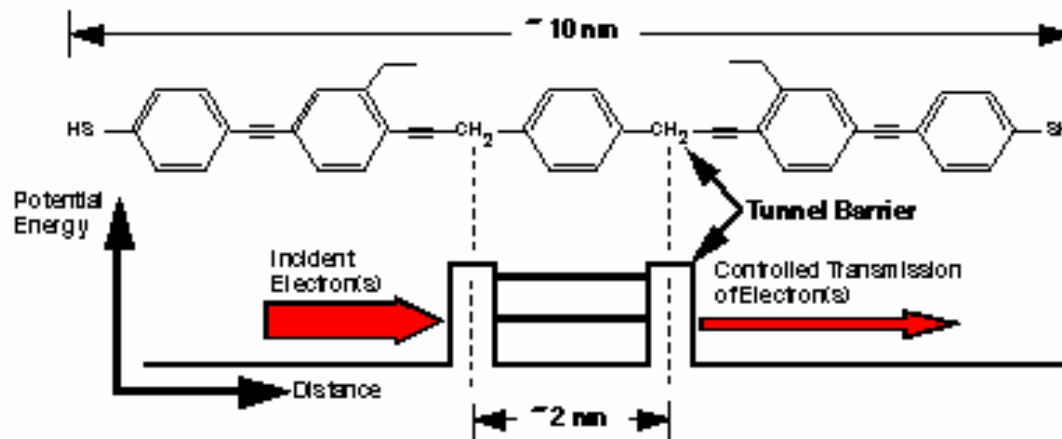
With a specific voltage applied, the E-field twists the molecule and permits current to flow.



# Molecular QM Devices

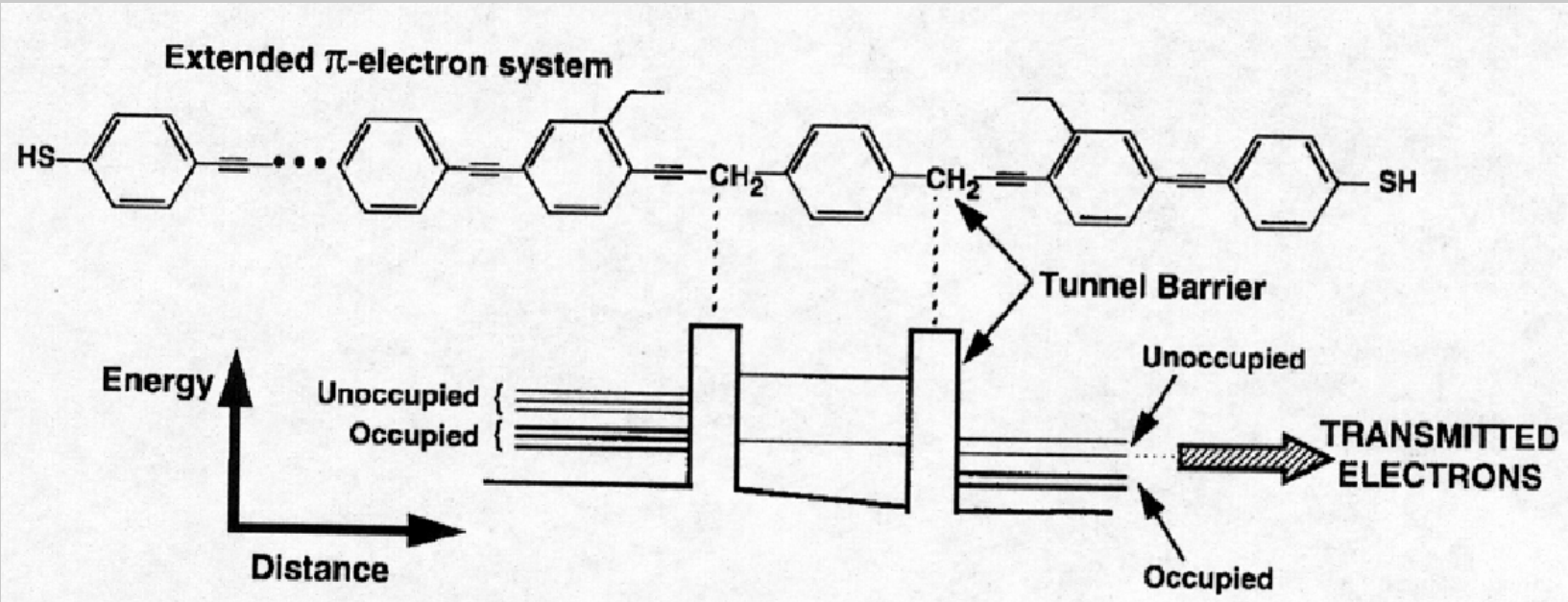
## Molecular Electronic Wires and Nanometer-Scale, Quantum-Effect Switches

- Molecule can act as wire or as resonant tunneling diode
- Methylene groups create “barriers” along a molecular wire to control transmission of electrons through a quantum well



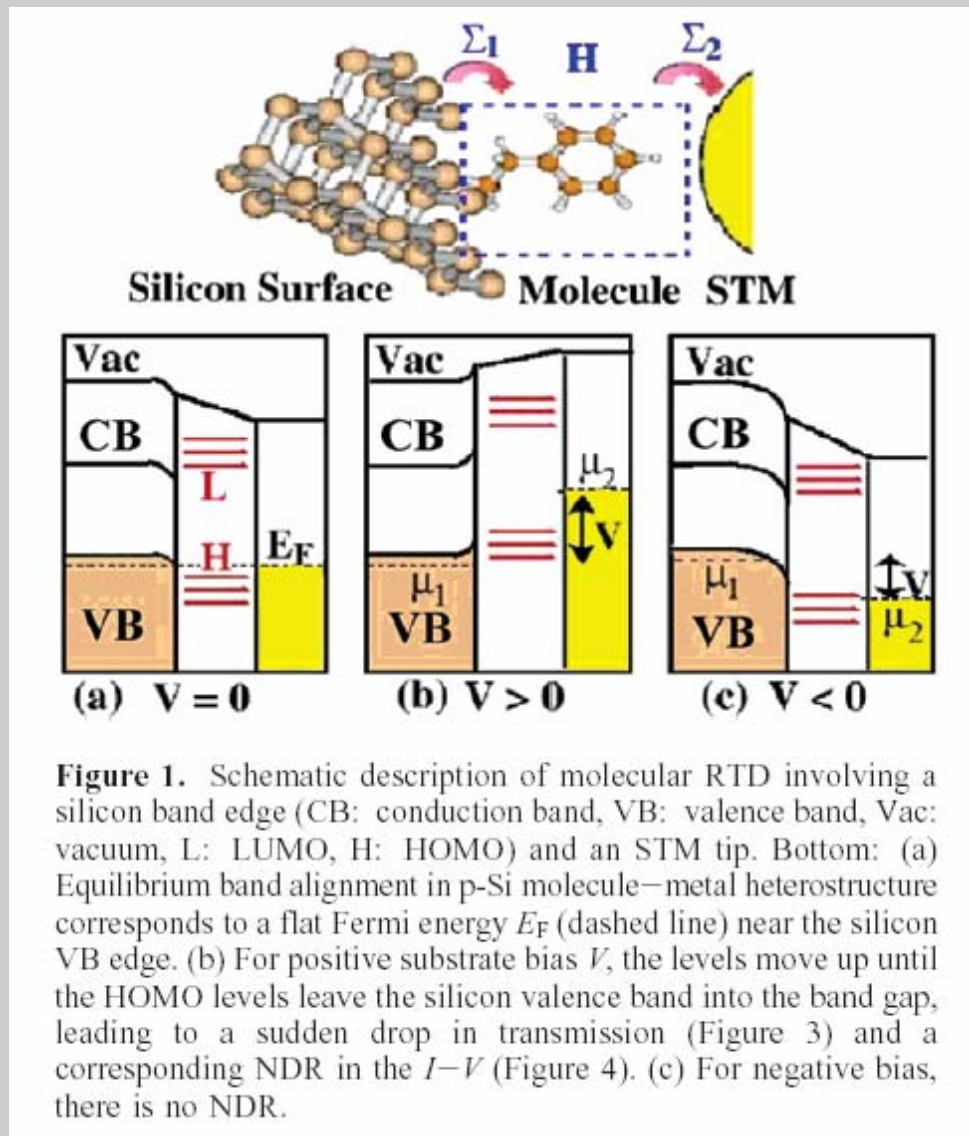
- Advantages:
  - Molecules much smaller and every one is exactly alike
  - Easily can be made in vast numbers ( $10^{23}$  at a time)

# Proposed Resonant Tunneling in a Conjugated Molecule

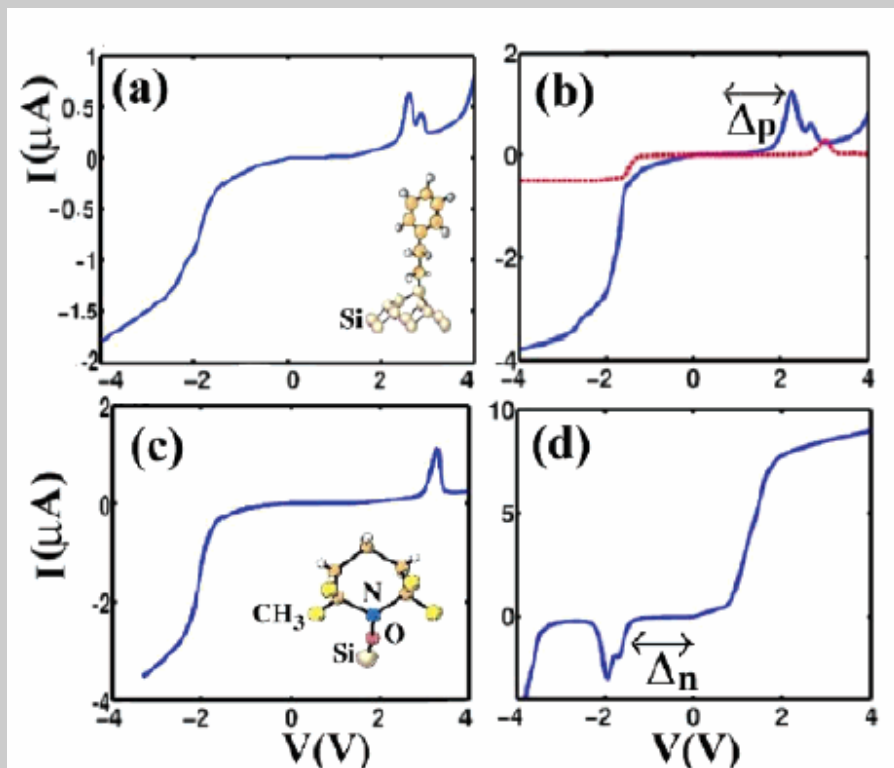


A potential problem is that it could be hard to line up energy levels in the source, island and drain since the source and drain have a limited number of energy levels.

# Silicon-Base Molecular Device - Theory

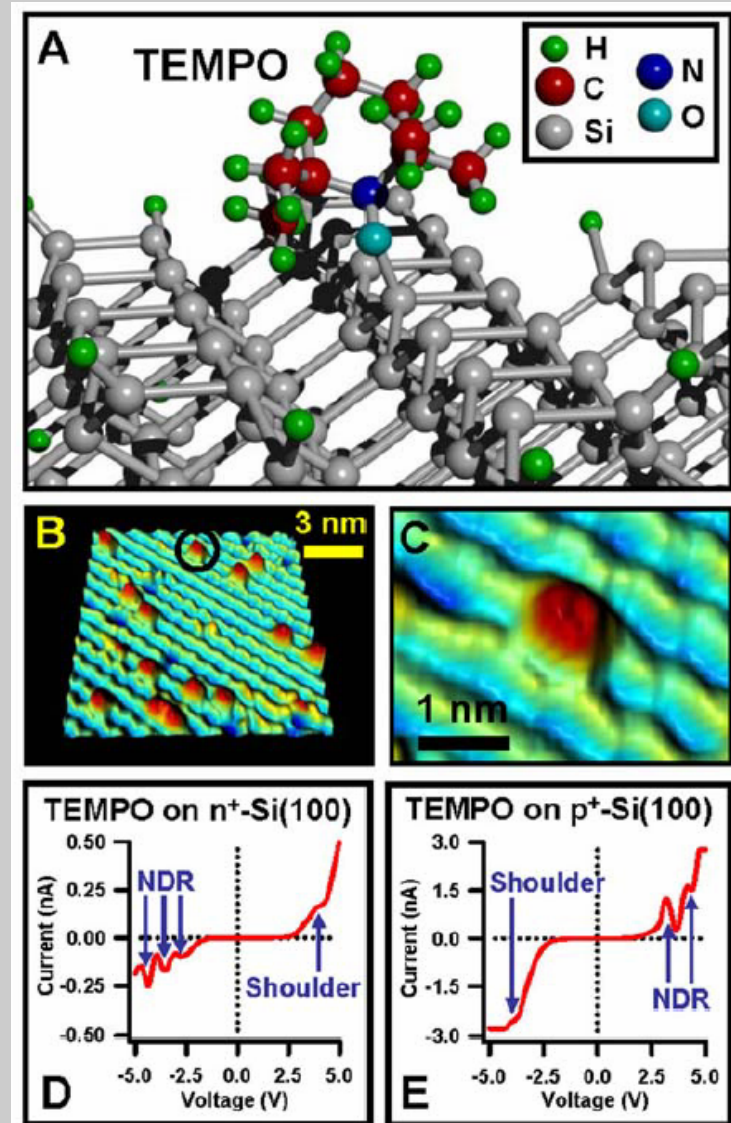
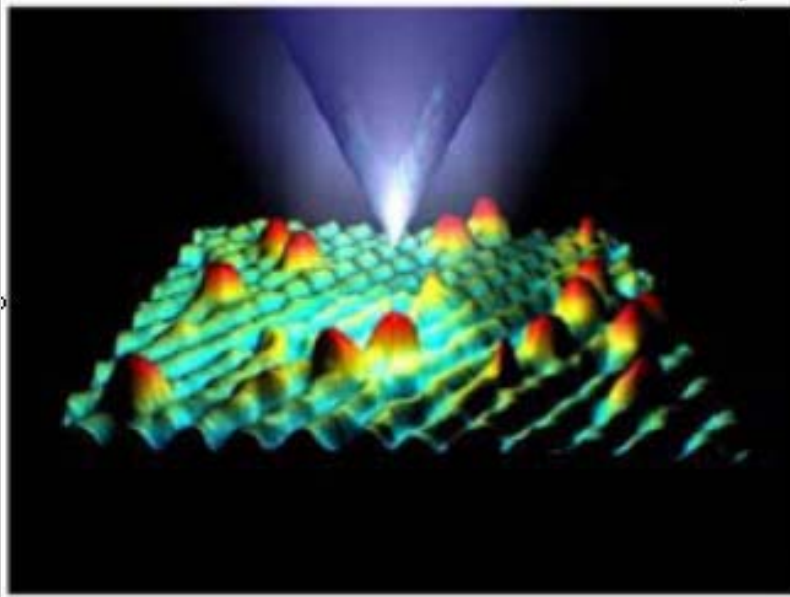


# Silicon-Base Molecular Device - Theory



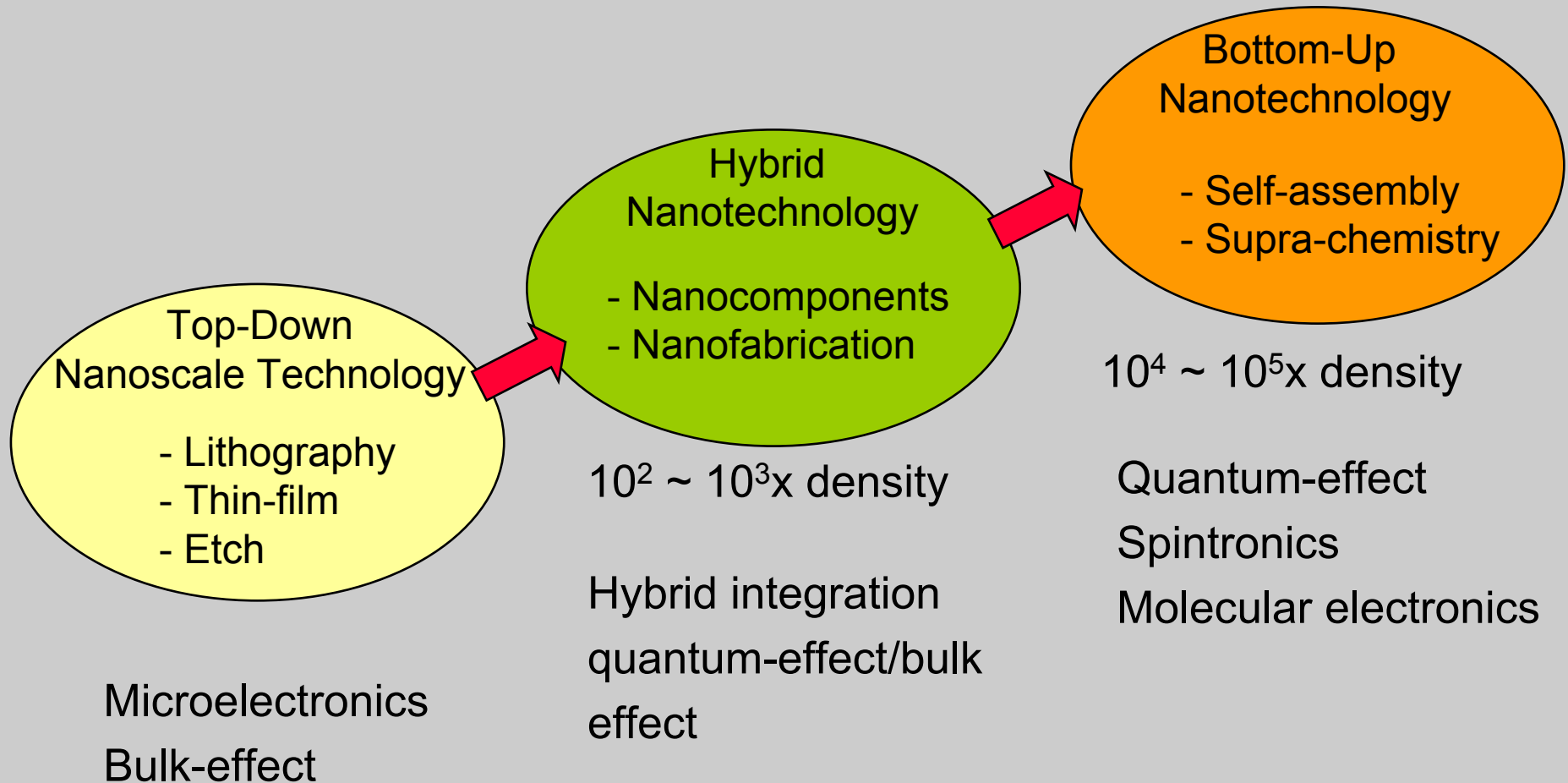
**Figure 4.** Calculated  $I$ - $V$  curves for styrene on p-doped H:Si(100) with an STM almost touching the molecule, using (a) EHT and (b) EM basis for Si. The calculations are non self-consistent at this time (ref 29). Increasing the air gap to 2 Å (dashed line) decreases the current and postpones the NDR. (c)  $I$ - $V$  for TEMPO on p-Si(100) (EM) and (d) styrene on n-doped H:Si(100) (EM) (ref 31). The NDR reverses polarity on reversing doping, although in calculations for this specific geometry, we had to lower the LUMO levels artificially to bring the NDR into the bias window.

# Silicon-Base Molecular Device - Experiment



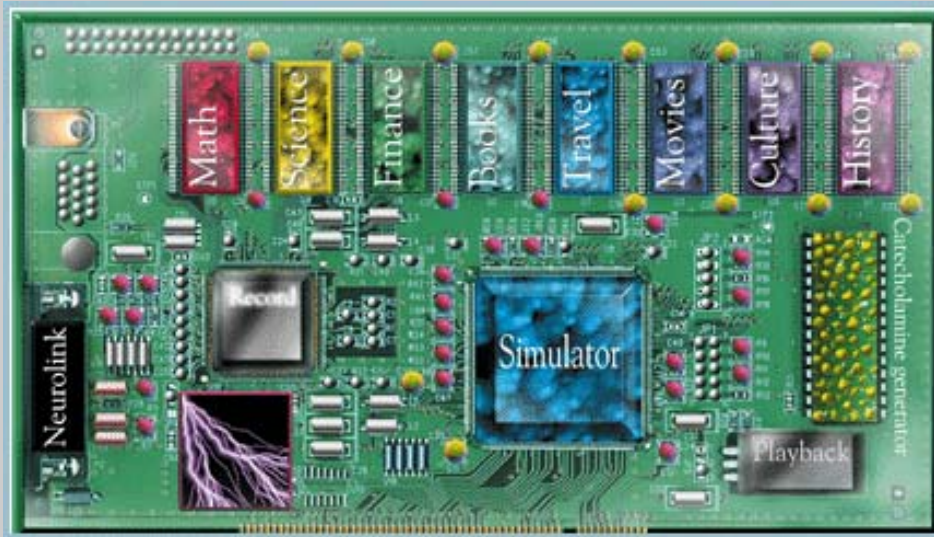
M. Hersan, 2004

# Electronics: Possible Evolution Path





# Merging of Mind and Machine



## Neural Implants

our intelligent creations will soon eclipse us

and that their creations will eventually eclipse them.

